STUDIES ON THE IONOPHOROUS ANTIBIOTICS. PART III. THE STRUCTURE OF LONOMYCIN, A POLYETHER ANTIBIOTIC

Noboru Otake and Mitsuo Koenuma

Institute of Applied Microbiology, The University of Tokyo Hiroshi Miyamae, Shoichi Sato and Yoshihiko Saito Institute for Solid State Physics, The University of Tokyo

Lonomycin¹⁾ is an antibiotic elaborated by *Streptomyces ribosidificus* and shows an antimicrobial activity against gram-positive bacteria including mycobacteria as well as some kind of filamentous fungi. Lonomycin is also effective in the treatment of coccidial infections in poultry and shows some chemical and biological characteristics common in the family of polyether antibictics.

(Received in Japan 17 September 1975; received in UK for publication 7 October 1975)

In the course of our serial studies²,³) on the ionophores, we have elucidated the entire structure of lonomycin as depicted in Fig. 1 using the thallium salt by a three dimensional X-ray analysis.

The thallium salt of lonomycin crystallizes from methanol-water in plates, m.p. $150.2-150.3^{\circ}$ C The molecular formula, C₄₄H₇₅O₁₄Tl is confirmed by comparison with the formulae of the methyl ester, C₄₅H₇₈O₁₄(M⁺-H₂O, m/e 824) and the sodium salt, C₄₄H₇₅O₁₄Na(M⁺, m/e 850). Crystal data: orthorhombic, space group P₂₁₂₁₂₁, a=16.257(2), b=25.731(4) and c=12.502(2)Å, Dm=1.36 (flotation in aqueous KI), Dc=1.33 g/cm³ for Z=4.

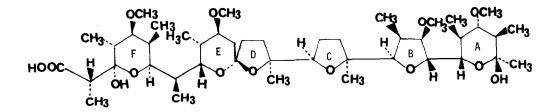


Fig. 1 The structure of lonomycin

4147

The intensity data were collected on an automated four-circle diffractometer using Mo Ka radiation (λ =0.7107Å). The maximum value of 20 was 53° in the ω -20 scan technique. Since the thallium salt of lonomycin decomposed gradually in the X-ray beam, four crystals were used and the intensities were corrected by comparison with the standard reflexions.

The structure was solved by the heavy-atom method. The positional and thermal parameters for the non-hydrogen atoms were refined by the block-diagonal least-squared method using anisotropic temperature factors. The final R-value for the 3225 reflexions used in the refinement was 0.078. The final coordinates and their standard deviations are given in Table 1.

Table 1. Positional parameters for the non-hydrogen atoms (× 10^3), with their e.s.d.'s in parentheses

4148

No. 47

The absolute configuration was determined by use of the anomalous dispersion effect of the thallium atom for Mo Ka radiation. Differences between intensities of some reflexions and those of their counter-reflexions were clearly discernible as shown in Table 2.

h k l	F _o (hkl)	F _c (hkl)	F _o (hkl)	F _c (hkl)
141	52	45	66	61
112	23	31	40	47
122	109	104	165	163
192	57	50	44	43
1 10 3	46	37	36	27
1 17 5	66	72	54	57
146	76	87	90	105
1 1 7	57	57	47	43
1117	83	80	70	70
1147	56	55	38	44
1 3 10	54	45	74	52

Table 2. Determination of absolute configuration of lonomycin

The resulting molecular structure of lonomycin thallium salt viewed along the c axis is illustrated in Fig. 2 which correctly represents the absolute configuration of the antibiotic. The whole molecule takes a circular conformation and the thallium (I) ion is located in a cavity The thallium atom co-ordinates to six oxygen atoms in the distance ranging from 2.6 to 3.0°_{A} .

The structure of lonomycin is similar to those found for nigericin⁴) and grisorixin⁵, however, it differs in the conformation of ring F and the side chain bearing the carboxylic function.

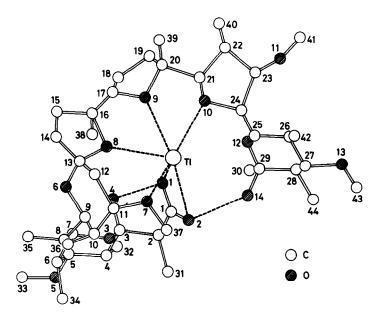


Fig. 2 The molecular structure of lonomycin viewed along the c-axis

Acknowledgement

We are grateful to Prof. H. Yonehara of the Institute of Applied Microbiology for his interest and discussions through this work and to Drs. S. Omura, J. Sawada and I. Tanaka of Taisho Pharmaceutical Co. Ltd. for the supply of lonomycin. This work was supported in part by a grant from the Ministry of Education, Science and Culture, the Government of Japan.

References

- 1) S. Omura, S. Machida, J. Sawada, I. Tanaka and N. Otake, Abstract Papers of the Annual Meeting of The Agricultural Chemical Society, Japan. page 84, Sapporo, 1975.
- 2) H. Kinashi, N. Ōtake, H. Yonehara, S. Sato and Y. Saito, Tetrahed. Letters 4955 (1973)
- 3) N. Otake, M. Koenuma, H. Kinashi, S. Sato and Y. Saito, Chem. Comm. 92 (1975)
- 4) L. K. Steinrauf, M. Pinkerton and J. W. Chamberlin, Biochem. Biophys. Res. Comm. <u>33</u> 29 (1968)
- 5) M. Alleaume and D. Hickel, Chem. Comm. 1422 (1970)